QUIZ: BIOGEOCHEMICAL CYCLES AND OZONE LAYER

1. Depletion of ozone layer is responsible for

- (a) Certain harmful radiations from the sun to reach the earth's surface.
- (b) Producing more oxygen in the earth's atmosphere.
- (c) Global Warming
- (d) Producing Greenhouse gases
- **2.** Carbon is incorporated into life forms directly through the basic process of photosynthesis using
 - (a) Carbonates in shells & water
 - (b) Petroleum
 - (c) Carbon dioxide
 - (d) Lime stone.
- **3.** The environmental consequences of increase in CO₂ level in the atmosphere causes
 - (a) Depletion of ozone layer
 - (b) Global warming
 - (c) Increase in acidity of soil
 - (d) Wild fire in forests
- **4.** Which of the following processes of water cycle is likely to cause drought in a specific area?
 - (a) Respiration
 - (b) Transpiration
 - (c) Greater rate of Evaporation
 - (d) Condensation
- **5.** Which one of the following examples is the gradual increase in earth's temperature?
 - (a) Depletion of ozone layer
 - (b) Increase in CO₂ level dissolved in oceans leading to higher temperature of ocean water
 - (c) Increase in CO₂ level in atmosphere leading to retention of more heat on earth
 - (d) Decrease in CO_2 level in the atmosphere thereby allowing more radiation from the sun to reach the earth.

- **6.** Oxides of nitrogen are produced in the atmosphere due to lightening. These nitrogen oxides are harmful to life forms and crops because
 - (a) They cause acid rain
 - (b) They cause respiratory problem
 - (c) They increase the percentage of nitrogen in the atmosphere
 - (d) They disturb the nitrogen cycle of nature

Answers:

1. (a)

Options:

- (a) Correct. As given in questionnaire sheet
- (b) Ozone depletion does not produce oxygen on the earth's surface
- (c) Global Warming is caused by increase of concentration of certain gases known as greenhouse gases
- (d) Greenhouse gases are not the result of ozone depletion but the human activities on the earth.

Explanation: Depletion of ozone layer causes ozone hole formation. As a result, harmful radiations from the sun reach the earth.

2. (c)

Options:

- (a) Carbon is found as carbonates and hydrogen carbonates in various minerals
- (b) Petroleum contains carbon in the form of compounds. It is one of major source of energy
- (c) Correct. As given in questionnaire sheet
- (d) Lime stone is not involved in photosynthesis.

Explanation: Carbon dioxide gets converted to glucose by all life forms that contain chlorophyll.

3. (b)

Options:

- (a) Depletion of Ozone layer is caused due to the presence of CFC's in the atmosphere and not by $CO_{2.}$
- (b) Correct. Same as in questionnaire sheet.
- (c) Soil becomes acidic due to acid rain which is caused due to the presence of gases of oxides of sulphur and nitrogen, in the atmosphere.
- (d) Increase in CO_2 level in the atmosphere will extinguish fire.

4. (c) Options:

- (a) Inhaling oxygen and exhaling carbon dioxide is respiration. It is not the cause of drought anywhere.
- (b) Releasing water in atmosphere by plants.
- (c) Correct. As given in questionnaire sheet.
- (d) Condensation of vapours helps in rainfall and not drought.

Explanation: The greater the rate of evaporation of water, greater is the chances for drought.

5. (c)

Options:

- (a) Ozone depletion causes holes in ozone layer leading to harmful radiations from the Sun to reach the earth.
- (b) Increase in CO₂ level dissolved in ocean does not cause increase of temperature of water as oceans are large sources of water.
- (c) Correct. Same as in questionnaire sheet.
- (d) More radiations from the sun reach the earth by ozone depletion and not by decrease of CO₂ level in the atmosphere.

Explanation: - Carbon dioxide being a greenhouse gas, does not allow the heat of the Earth to escape into the atmosphere.

6. (a)

Options:

- (a) Correct. Same as in questionnaire sheet.
- (b) Nitrogen oxides do not cause respiratory problems to crops.
- (c) The nitrogen in the atmosphere becomes a part of nitrogen cycle.
- (d) Nitrogen cycle is not disturbed as nitrogen oxides are converted to nitrogen containing molecules.